

<p style="text-align: center;">Warren County School District</p> <p style="text-align: center;">PLANNED INSTRUCTION</p>

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COURSE DESCRIPTION

Course Title: Advanced Chemistry (Honors)

Course Number: 00333

Course Prerequisites: Successful Completion of Academic Chemistry or permission of the principal

Course Description: (Include “no final exam” or “final exam required”)

Advanced Chemistry provides able and motivated students with the opportunity to pursue college-level chemistry studies while still in high school. This rigorous preparatory course is designed for students who will study chemistry or a related field at the college level. Advanced Chemistry is a one-year, one-credit honors course. Topics include: crystallography, ideal gases, nuclear chemistry, thermodynamics, quantum mechanics, bonding and coordination chemistry, as well as career exploration within the field of chemistry.

Suggested Grade Level: 12

Length of Course: _____ One Semester X Two Semesters _____ Other
(Describe)

Units of Credit: 1 (Insert **NONE** if appropriate.)

PDE Certification and Staffing Policies and Guidelines (CSPG) Required Teacher Certification
(Insert certificate title and CSPG#)

Certification verified by WUSD Human Resources Department:
Yes No

Board Approved Textbooks, Software, Materials:

Title: _____

Publisher:

ISBN #:

Copyright Date:

Date of WCSD Board Approval:

BOARD APPROVAL:

Date Written: November 2009

Date Approved: _____

Implementation Year: 2010-2011

Suggested Supplemental Materials:

Course Standards

PA Academic Standards: (List by Number and Description)

3.1 Unifying Themes

12B Apply concepts of models as a method to predict and understand science and technology.

12C Assess and apply patterns in science and technology.

3.2 Inquiry and Design

12B Evaluate experimental information for appropriateness and adherence to relevant science processes.

12C Apply the elements of scientific inquiry to solve multi-step problems.

3.4 Physical Sciences, Chemistry and Physics

12A Apply concepts about the structure and properties of matter.

12B Apply and analyze energy sources and conversions and their relationship to heat and temperature.

4.3 Environmental Health

12A Analyze the complexity of environmental health issues.

12B Analyze the local, regional and national impacts of environmental health.

WCSD Academic Standards: (List or **None**)

None

Industry or Other Standards: (List, Identify Source or **None**)

None

WCSD EXPECTATIONS

WCSD K-12 Expectations for instruction in writing, reading, mathematics and, technology have been developed and revised annually. The teacher will integrate all WCSD Expectations into this planned instruction.

SPECIAL EDUCATION AND GIFTED REQUIREMENTS

The teacher shall make appropriate modifications to instruction and assessment based on a student's Individual Education Plan (IEP) or Gifted Individual Education Plan (GIEP).

SPECIFIC EDUCATIONAL OBJECTIVES/CORRESPONDING STANDARDS AND ELIGIBLE CONTENT WHERE APPLICABLE

(List Objectives, PA Standards #'s, Other Standards (see samples at end))

This course is written to the 12th grade standards. No assessment anchors have been written for this level.

ASSESSMENTS

PSSA Assessment Anchors Addressed: The teacher must be knowledgeable of the PDE Assessment Anchors and/or Eligible Content and incorporate them into this planned instruction. Current assessment anchors can be found at pde@state.pa.us.

No assessment anchors are available for this course. See above.

Formative Assessments: The teacher will develop and use standards-based assessments throughout the course.

Portfolio Assessment: _____ Yes X No

District-wide Final Examination Required: _____ Yes X No

Course Challenge Assessment (Describe):

REQUIRED COURSE SEQUENCE AND TIMELINE

(Content must be tied to objectives)

Content Sequence	Dates
A. Gases 15 days <ul style="list-style-type: none">1. Avogadro's Hypothesis2. Molar volume3. Ideal Gas Law4. Gas reaction Stoichiometry	
B. Limiting reactants 10 days	
C. Solids 20 days <ul style="list-style-type: none">1. Crystal systems2. Unit cells3. Closest packing4. Semiconductors5. Liquid crystals6. Amorphous substances7. Hydrates	
D. Nuclear chemistry 20 days	

1. Types of radiation
3. Elementary particles
4. Quark theory
5. Balancing nuclear equations
6. Decay and half-life
7. Nuclear reactions
 - a. Fission
 - b. Fusion

E. Thermodynamics 20 days

1. Enthalpy
2. Entropy
3. Free energy

F. Oxidation-reduction 20 days

1. Oxidation
2. Reduction
3. Balancing Redox equations

G. Electrochemistry 20 days

1. Electrolytic conduction
2. Metallic conduction
3. Electrolysis
4. Voltaic cells
5. Nernst equation

H. Quantum-mechanical model of the atom 15 days

1. Bohr atom
2. De Broglie's Hypothesis
3. Heisenberg Uncertainty Principle
4. Schrodinger's Wave Equation
5. Quantum numbers
6. Pauli Exclusion Principle

I. Molecular structure 20 days

1. VSEPR
2. Atomic orbital overlap
3. Hybridization
4. Resonance
5. Molecular orbitals

J. Coordination Chemistry 20 days

1. Nomenclature
2. Coordinate covalent bonds
3. Ligand Field Theory

K. Laboratory Time: Throughout course

Objectives:

1. Evaluate scientific processes by collecting data and applying knowledge to physical models to interpret data.
2. Assess and apply patterns in equilibrium systems, nuclear reactions, and the structure of matter.
3. Evaluate the atomic theory.
4. Evaluate data correctly for conclusions.
5. Characterize matter in terms of its structure and properties.

WRITING TEAM: Sue Hutchins, Charles Hayes

WCSD STUDENT DATA SYSTEM INFORMATION

1. Is there a required final examination? ☒ Yes ☐ No
2. Does this course issue a mark/grade for the report card?
☒ Yes ☐ No
3. Does this course issue a Pass/Fail mark? ☐ Yes ☒ No
4. Is the course mark/grade part of the GPA calculation?
☒ Yes ☐ No
5. Is the course eligible for Honor Roll calculation? ☒ Yes ☐ No
6. What is the academic weight of the course?
☐ No weight/Non credit ☐ Standard weight
☒ Enhanced weight (Honors) As per current school board policy.